

Review:

$$\text{pOH} = 3.71$$

$$\text{pH} =$$

$$[\text{H}_3\text{O}^{+1}] =$$

$$[\text{OH}^{-1}] =$$

Write the hydrolysis reaction of HBr

Write the hydrolysis reaction of NH_3 .

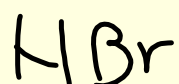
Review:

$$\text{pOH} = 3.71$$

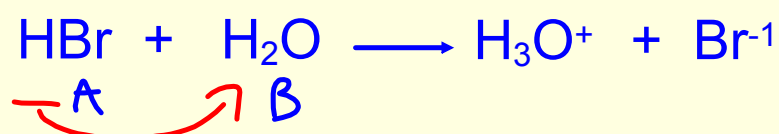
$$\text{pH} = 10.29$$

$$[\text{H}_3\text{O}^{+1}] = 5.13 \times 10^{-11}$$

$$[\text{OH}^{-1}] = 1.95 \times 10^{-4}$$



Write the hydrolysis reaction of HBr



Write the hydrolysis reaction of NH₃.

